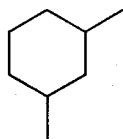


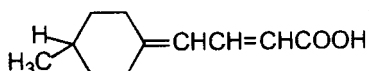
NUMBER OF STEREOISOMERS

Q.1 How many stereoisomers are possible for the following molecule ?



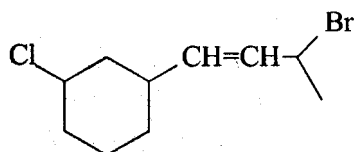
- (A) 1 (B) 2 (C) 3 (D) 4

Q.2 How many stereoisomers are possible for the following molecule ?



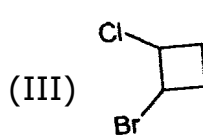
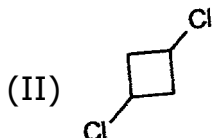
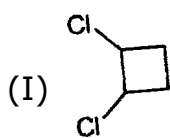
- (A) 1 (B) 2 (C) 3 (D) 4

Q.3 How many stereoisomers are possible for the following molecule :



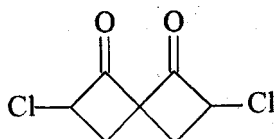
- (A) 4 (B) 8 (C) 10 (D) 16

Q.4 What is the correct order of compounds for increasing number of stereoisomers



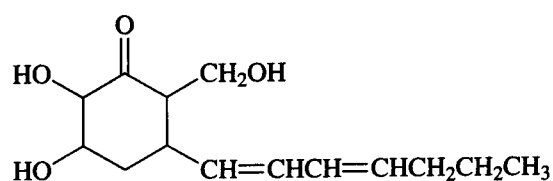
- (A) II < I < III (B) I < II < III
 (C) II < III < I (D) All have same no. of isomers.

Q.5 How many stereoisomers are possible for the following molecule ?



- (A) 2 (B) 3 (C) 4 (D) 8

Q.6 A naturally occurring substance has the constitution shown. How many stereoisomers may have this constitution ?



- (A) 2 (B) 8 (C) 16 (D) 64

Q.7 If a compound has n asymmetric carbon atoms, then maximum number of optical isomer are given by the formula

- (A) $\left(\frac{1}{2}\right)^n$ (B) 2^n (C) $2\sqrt{n}$ (D) $\sqrt{2n}$

Q.8 The number of possible racemic forms in Glucose are -

- (A) 4 (B) 8 (C) 12 (D) 16

Q.9 How many stereoisomeric aldohexoses are possible ? (All are known).

- (A) 4 (B) 8 (C) 12 (D) 16

Q.10 The number of meso forms in the compound given below is
 $\text{HOOCCH}(\text{CH}_3)\text{CH}(\text{OH})\text{CH}(\text{Cl})\text{CH}(\text{OH})\text{CH}(\text{CH}_3)\text{COOH}$

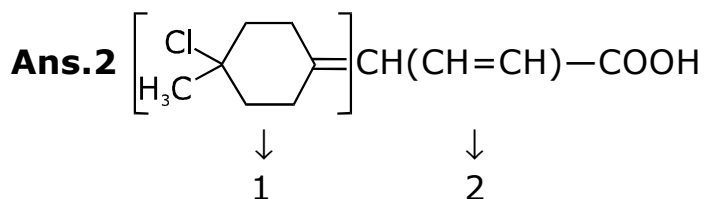
- (A) 4 (B) 3 (C) 16 (D) 8

ANSWER KEY

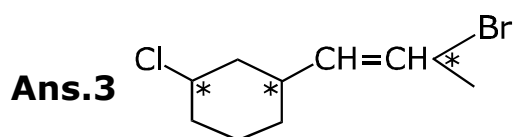
Q.No.	1	2	3	4	5	6	7	8	9	10
Ans.	C	D	D	A	D	D	B	B	D	A

SOLUTIONS (NUMBER OF STEREOISOMERS)

Ans.1 Molecule has an even no. of chiral carbon atoms and has an plane of symmetry
 optically active forms = 2^{n-1}
 $n = \text{no. of chiral carbons} = 2$
 $= 2^{2-1} = 2$
 Meso form = $2^{n/2-1} = 2^{2/2-1} = 1$
 Total no. of stereoisomers = $2 + 1 = 3$

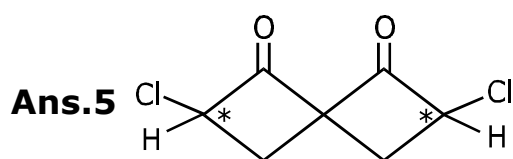


There are two stereogenic centre in above molecule
 $2^2 \Rightarrow 4$

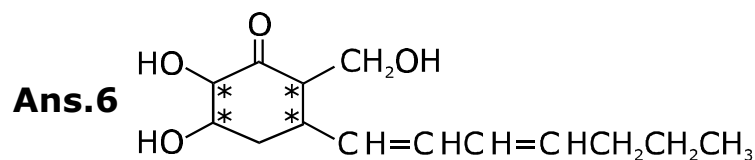


Total number of chiral centers are 4 (one due to one double bond)
 No. of stereoisomers = 2^2
 $n = \text{No. of stereogenic centre}$
 $= 2^4 \Rightarrow 16$

Ans.4 II has no chiral carbon.
 III has even no. of chiral carbons (i.e. 2) and has no plane of symmetry.
 Optically active forms = $2^2 = 4$
 Meso = 0
 Total no. of stereoisomers = 4
 I has 2 chiral carbon & has plane of symmetry
 optically active forms = $2^{2-1} = 2$
 Meso = $2^{2/2-1} = 1$
 Total no. of stereoisomers = $2 + 1 = 3$



Molecule has 3 stereogenic centre i.e. 2 are chiral carbon atoms (marked with asterisk) and one stereogenic centre is due to spiro component.
 Total no. of stereoisomers = $2^3 = 8$



$$2^6 \Rightarrow 64$$

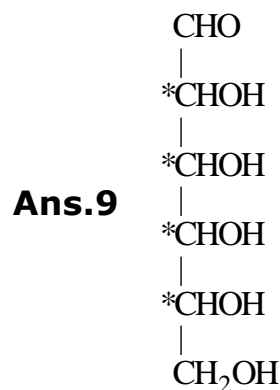
There are 6 stereogenic centres.

(4 due to four chiral centres and 2 due to two double bonds)

Ans.7 2^n

Ans.8 Glucose has 4 chiral carbons and it is unsymmetrical molecule so

$$\text{number of racemic forms are } \frac{2^n}{2} = \frac{2^4}{2} = 8$$



Aldohexose

There are 4 chiral carbon atoms in aldohexose

$$\begin{aligned} \text{No. of stereoisomers} &= 2^4 \\ &= 16 \end{aligned}$$

Ans.10 The number of chiral carbons = 5 (odd number). It is symmetrical molecule so no. of meso forms

$$= 2^{\binom{n-1}{2}} = 2^{\binom{5-1}{2}} = 2^2 = 4.$$